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(i)	P	rinted Page	es :4]	Roll No		•••••	•••••	•••••	
(ii)	Q	uestions	:9]	Sub. Code:	0	1	5	0	
				Exam. Code:	0	0	0	2	
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		Paper -	- VI : C	rganic Cher	nist	ry-l	В		
Time	e:3	Hours]			[1	Max.	Mark	ks:45)
		each Section carry equa		tion No. IX is comp	ulsory.	All c	quest	ions	
				SECTION-I					
l. (á	a)	Bromine is less reactive but more selective whereas chlor is more reactive but less selective. Explain with example.							
	(b)	Define Wu	rtz-reacti	on. Discuss its mech	nanism	•		3	
	(c)	Explain Sac	chse-Mol	nr theory of strain le	ess ring	S.		3	
II.	(a)	What is Co Wurtz reac	•	e reaction ? Discuss	its adv	/anta	ages	over 3	
	(b)			nism of chlorination support of the abov				d give 3	
	(c)	Discuss Bay	yer's Stra	in theory and its lim	nitation	ıs		3	

SECTION-II

III. (a) What do you mean by Saytzeff's Rule? Give one example. arrange the following alkenes in order of their stabilities but-2-ene, 2-methyl but-2-ene, 2,3-dimethyl but-2-ene.

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- (b) Discus the mechanism of Anti Markownikoff's addition to alkenes.
- (c) Complete the following:

(i)
$$CH_3 - CH = CH_2$$
 (ii) B_2H_6 (ii) H_2O_2 OH

(ii)
$$CH_3 - CHOH - CH_3$$
 Conc. H_2SO_4

(iii)
$$CH_3 - CH = CH - CH_3 + CH_2N_2 \xrightarrow{hv}$$

- IV. (a) Give the mechanism of oxymercuration reduction. 3
 - (b) What do you understand by 1,2-hydride shift? Explain with suitable examples from dehydration of alcohols.
 - (c) Write note on ozonolysis olefines.

SECTION-III

- V. (a) What is the cause of acidic nature of alkynes? Explain 3
 - (b) Which is more stable penta-1,4-diene or buta-1,3-diene? Explain

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(c) Complete the following

(i)
$$CH = CH + H_2O \xrightarrow{H_2SO_4/HgSO_4}$$

(ii)
$$CH = CH + O_3$$
 $\xrightarrow{Zn/CH_3CHOH}$

(iii)
$$CH = CH$$
 $CuCI/NH_4CI$

- VI. (a) What is Diels Aloer reaction referred to as (4 + 2) Cycloadditlon reaction? Explain with example.
 - (b) Give any three methods for the preparation of dienes. Explain why alkynes undergo nucleophilic addition reactions.

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SECTION-IV

- VII. (a) Chlorine deactivates benzene ring but it isortho, para directing. Justify.
 - (b) Define:
 - (i) Huckel's Rule
 - (ii) Annulenee.
 - (c) Write short notes on
 - (i) α and π complexes
 - (ii) Birch reduction.
- VIII. (a) What are aromatic, antiaromatic and non-aromatic compounds? Give examples.
 - (b) Define Friedal Craft's reaction. Give its mechanisms.

SECTION-V(Compulsory)

- IX. (i) Why are cycloalkanes with large rings not easily formed?
 - (ii) What is meant by orientation in halogenation of alkanes?
 - (iii) What is Epoxidation?
 - (iv) Explain the term torsional strain.
 - (v) Define the term dienophile.
 - (vi) How can you convert propyne to 2-pentyne?
 - (vii) Give molecular orbital structure of benzene.
 - (viii) Identify the product A and B:

(xi) Give one method of formation of biphenyl. 1x9=9