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**Sub Code**: 0547(1048) **Exam Code**: 0006

**Exam**: B.A./B.Sc. (General), 6th Semester

**Subject:** Chemistry

Paper: Paper-XXI: Inorganic Chemistry-B

Time: 3 Hours Maximum Marks: 22

**Note:** Attempt five questions in all, selecting one question from each Unit. Unit-V is compulsory.

# **UNIT-I**

 (a) What is Polymeric backbone in silicone and Phosphazenes?

2

2

2

(b) What are Homomorphic and Heteromorphic T1- system. Explain with examples.

2. (a) How is Cyclic (NPCI<sub>2</sub>)<sub>3</sub> prepared? Give an account of its Nucleophilic substitution reactions.

(b) What are silicone oils and silicone Rubbers. Explain with one example each.

### **UNIT-II**

3. (a) Define HSAB Principle. Discuss the applications and limitations of this Principle.

2

(b) What is relationship between Electronegativity and Hardness. Give example.

2

4. (a) What lead to hard - hard and soft - soft interactions. Give evidences in support. What are its consequences?

2

(b) How will you determine the relative Strength of Hard and Soft Acids. and Bases?

2

## **UNIT-III**

5. (a) What are Orgel Diagrams? Draw Orgel energy evel diagram for d¹ configuration in tetrahedral system. What are the limitations of these diagrams.?

2

(b) State and explain two selection rules for Electronic Absorption spectrum of transition metal complexes.

2

6. (a) Why do tetrahedral complexes of transition metal elements give much more intense d-d electronic spectra than in octahedral complexes?

(b) How Russell Saunders States get splitted In octahedral fields? Explain with diagram.

2

### **UNIT-IV**

7. (a) What is Origin of magnetism? How is magnetic susceptibility measured?

2

(b) Calculate spin only magnetic moment for :

2

- (i)  $Ni^{2+}$
- (ii) Cu<sup>2+</sup>
- (iii) Co<sup>3+</sup>
- (iv) Cr<sup>2+</sup>

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- 8. (a) Write short notes on the following:
  - (i) Diamagnetic Correction
  - (ii) Ferromagnetism and Antiferromagetism
  - (b) Explain spin and Orbital contribution to magnetic mements. Give example.

2

# UNIT-V Compulsory Questions

6

- 9. (a) What is Absolute Hardness.
  - (b) What is Curie's Point and Neel Temperature
  - (c) Write Mulliken symbol for spectroscopic terms D and F in Tetrahedral fields.
  - (d) Define soft base with one example.
  - (e) Name one ion which shows Temperature independent Paramagnetism.
  - (f) Give one important use of silicone Rubber